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Collision Theory Model: Collision Theory In the picture below, the baseball bat represents Reactant A and the baseball represents Reactant B A reaction will only be successful if the batter hits a homerun. If the batter does not hit a homerun, the reaction will be considered a failure. Now, read the four scenarios below and answer the key questions that follow Scenario 1: The pitcher throws a ...

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Based on your answer to Key Question 5, complete the following statement: Collision theory states that a reaction is most likely to occur if... If the reactants collide with the correct amount of energy and correct orientation. Exercise 1. Hydrogen gas and iodine vapor combine to form hydrogen iodide gas, as shown in the equation $H_2 + I_2 \rightarrow 2 HI$.

[Collision Theory - Impact for a Chemical Reaction](#)

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The collision theory states that a chemical reaction can only occur between particles when they collide (hit each other). The collision between reactant particles is necessary but not sufficient for a reaction to take place. The collisions also have to be effective.

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Collision Theory Pogil Answers Key Based on your answer to Key Question 5, complete the following statement: Collision theory states that a reaction is most likely to occur if… If the reactants collide with the correct amount of energy and correct orientation.

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Collision Theory. Collision theory basically explains how reactions occur and why different reactions have different reactions rates. It states that: Molecules must collide in order to react. In order to effectively initiate a reaction, the molecules in the collisions must have sufficient energy to bring about disruptions in the bonds of molecules.

[Mrs. Zuberbuehler - Mrs. Zuberbuehler](#)

©POGIL - 2005 Authored by LIQUID Edited by Linda Padwa and David Hanson, Stony Brook University . Kinetic Molecular Theory 2. In which phase of matter is there the least spacing between particles? 3. In which phase of matter is there the most potential for movement? 4. Which phase of matter does not have a definite shape yet the particles will not fill the container? 5. In terms of spacing ...

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Collision theory states that a reaction is most likely to occur if and OPOGIL 2005, 2006 Authored by: Bryan Horan; Revised by: Kelly Levy and Kenneth Levy Edited by Linda Padwa and David Hanson, Stony Brook University 3/4 . Collision Theory 7. With your group, develop a different analogy/model to explain the collision theory to someone who is not in your group. Exercise 1. Hydrogen gas and ...

